LESSON PLAN (ODD SEMESTER 2025-26)

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B.Sc. 1st Year (SEC-Chemistry)

Month/Week	First week	Second week	Third week	Fourth week
August	Composition of	Complexometric titrations,	Use of indicators, estimation of	Pure
	soil, concept of pH and pH	chelation,	calcium and	water, sources
	measurement of	<u>-</u>	magnesium ions	responsible for contaminating
	soil,	chelating agents,	in soil.	water, water
	3011,		111 3011.	sampling
				methods,
				methous,
September	Water	A general study	A general study	A general study
	purification	including	including	including
	methods,	preparation and	preparation and	preparation and
	determination of	uses of the	uses of the	uses of the
	dissolved oxygen	following: Hair	following: suntan	following:
	of a water	dye, soap,	lotions,	lipsticks, talcum
	sample.	shampoo	face powder	powder, nail
				enamel.
October	General	Synthesis and	Revision	Basic principle of
	introduction to	technical		pH metric,
	pesticides	manufacture and		potentiometric
	(natural and	uses of		and
	synthetic),	representative		conductometric
	benefits and	pesticides in the		titrations,
	adverse effects,	following classes:		
	changing	organochlorines		
	concepts	(gammexene),		
	of pesticides,	organophosphates		
November	Applications of	(malathion). Buffer	Dovision	Davision
ivovember	Applications of conductivity	solution, buffer	Revision	Revision
	measurements:	action,		
	determination of	Henderson–Hazel		
	degree of	equation, buffer		
	dissociation,	mechanism of		
	determination of	buffer action.		
	Ka of acids and	Sairer action.		
	base,			
	vase,	1	l	1

B.Sc. II Year (SEC-Chemistry)

Month/Week	First week	Second week	Third week	Fourth week
August	General introduction, designing to eliminate potential safety problems, battery safeguards when using discrete batteries, battery construction,	Design of rechargeable batteries, factors affecting battery performance.	General characteristics and applications of primary batteries,	Types and characteristics of primary batteries,
September	Comparison of the performance characteristics of primary battery systems, recharging primary batteries.	Zinc-Carbon Batteries (Leclanche´ and Zinc Chloride Cell Systems)	General characteristics, cell chemistry, types of cells and batteries, construction, cell components.	Magnesium and Aluminum Batteries:
October	Mg/MnO2 batteries, other types of magnesium primary batteries.	General characteristics and applications of secondary batteries, types and characteristics of secondary batteries,	Revision	Comparison of performance characteristics for secondary battery systems and introduction, chemistry, construction, performance characteristics,
November	Leadbatteries, Lithium ionbatteries, Iron electrode batteries,	Nickel-Cadmium, Nickel-Metal hydride, Nickel- Zinc batteries.	Revision	Revision

B.Sc. III Year (Inorganic Chemistry)

Month/Week	First week	Second week	Third week	Fourth week
August	Limitations of valence bond theory, an elementary idea of crystal-f ield theory,	crystal field split ting in octahedral	crystal field split ting in tetrahedral and square planar complexes	factors affecting the crystal-f ield parameters.
September	A brief outline of thermodynamic stability of metal complexes and factors affec ting the s tab ility,	substitution reac tions of square planar complexes of Pt(II).	Types of magnetic behaviour, methods of determining magnetic susceptibility, spin-only formula.	L-S coupling, correlation of s and eff values, orbital contribution to magnetic moments,
October	application of magnetic moment data for 3dmetal complexes.	Types of electronic transitions, selection rules for d-d transitions, spectroscopic ground states, spectrochemical series.	Revision	discussion of the electronic spectrum of [Ti(H2O)6]3+ complex ion.
November	Orgel-energy level diagram for d1 and d9 states	Revision	Revision	Revision

B.Sc. III Year (Organic Chemistry)

Month/Week	First week	Second week	Third week	Fourth week
August	Principle of	equivalent and	proton	Discuss ion of
	nuclear magnetic	nonequivalent	counting,splitting	PMR spectra of
	resonance, the	protons positions	of signals and	the molecules
	PMR	of signals and	coupling	
	spectrum,number	chemical	constants,	
	of signals, peak	shift, shielding and	magnetic	
	areas,	deshielding of	equivalence of	
		protons,	protons.	
September	Classification and	Configuration of	Formation of	Mechanism
	nomenclature.	monosaccharides.	glycos ides,	ofmutarotation.
	Monosaccharides,	Erythro and threo	ethers and esters.	Structures of
	mechanism of	diastereomers.	Determination of	ribose and
	osazone	Conversion of	ring size of	deoxyribose.
	formation, inte	glucose in to	glucose and	
	rconversion of	mannose.	fructose. Open	
	glucose and		chain and cyclic	
	fructose, chain		structure of D(+)-	
	lengthening and		glucose & D(-)	
	chain shortening		fructose.	
	of aldoses.			
October	An introduc tion	Organomagnesium	Revision	Organolithium
	to disaccharides	compounds: the		compounds:
	(maltose, sucrose	Grignard reagents-		formation and
	and lactose) and	formation,		chemical
	polysaccharides	structure and		reactions.
	(starch and	chemical		
	cellulose) without	reactions. O		
	involving			
	structure			
	determination.			
November	Organozinc	Revision	Revision	Revision
	compounds:			
	formation and			
	chemical			
	reactions.			